## **Centerband-Only Detection of Exchange: Efficient** Analysis of Dynamics in Solids by NMR

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Molecular dynamics have important effects on mechanical and conduction properties of polymers,<sup>1,2</sup> activity of proteins,<sup>3</sup> stability of pharmaceuticals,<sup>4</sup> transport properties in zeolites,<sup>5</sup> and behavior of amorphous materials near the glass transition.<sup>6</sup> Solid-state NMR provides powerful techniques for characterizing such dynamics in detail, in particular through multidimensional exchange experiments.<sup>7</sup> So far, however, the sensitivity and resolution of most of these NMR techniques has been limited, since anisotropybroadened line shapes or strong sidebands in magic-angle spinning spectra (MAS) were required. In this paper, we describe a new NMR pulse sequence for observing and characterizing slow (k = 0.2/s to 2000/s) segmental reorientations with the highest available NMR sensitivity and site resolution, in sideband-free <sup>13</sup>C, <sup>29</sup>Si, <sup>31</sup>P, or <sup>15</sup>N MAS spectra. It is termed *c*enterband-*o*nly detection of exchange, CODEX. From two short series of onedimensional MAS spectra, the correlation function, correlation time, and motional amplitude can be determined for each site with a resolved line in the MAS spectrum.

The pulse sequence for the CODEX experiment is shown in Figure 1. For N/2 rotation periods, the spins evolve under the anisotropic chemical shift, which is recoupled<sup>8</sup> by two 180° pulses per rotation period  $t_r$  in a spinning-speed independent fashion.<sup>9</sup> Then, the magnetization is stored along the z-direction, so that it does not precess or dephase during the long mixing time  $t_m$ , which is an integer multiple of  $t_r$ . If no motions occurred during  $t_m$ , the chemical-shift evolution, after a read-out pulse and another N/2rotation period under the recoupled chemical-shift anisotropy, is refocused at the start of detection. If segmental reorientation did occur during  $t_{\rm m}$ , the orientation-dependent frequency has changed and the chemical-shift anisotropy is not completely refocused. The dephasing is observed with high sensitivity as a decrease in the detected line intensity. High spinning rates  $v_r$  can be used to obtain virtually sideband-free spectra, since the dephasing is only a function of the total time  $Nt_r$ .

The often dominant signal of immobile sites can be removed by subtracting the CODEX spectrum from a reference spectrum, without an adjustable scaling factor. This reference spectrum is obtained simply by interchanging  $t_m$  and  $\Delta$  in the pulse sequence

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Figure 1. Pulse sequence for CODEX NMR: 90° pulses are filled black, 180° pulses white. CP: Cross polarization, DD: dipolar decoupling. The basic phase cycle of the 90° pulses is indicated, to be complemented by phase cycles to remove artifacts from  $T_1$  relaxation during  $t_m$  and  $\Delta$ . TOSS (total suppression of spinning sidebands) can be applied before detection with incremented rotor phase  $kt_r/K$ . Pure-exchange CODEX spectra are obtained by measuring a reference spectrum with  $t_{\rm m}$  and  $\Delta$  interchanged and subtracting the CODEX spectrum from it. Measurements are alternated between CODEX and reference spectrum every 5 min to eliminate the effects of spectrometer drift.



Figure 2. (a) CODEX of unlabeled methylmalonic acid at  $v_r = 5.5$  kHz. Middle: CODEX spectrum with  $t_m = 200$  ms and  $\Delta = 1$  ms. Top: Reference spectrum with  $\Delta = 200$  ms and  $t_m = 1$  ms. Bottom: The difference between the top and middle (the pure-exchange CODEX spectrum). For the COO carbon,  $\delta N t_r = 7.6 \pi$ . "s.s.": spinning sideband. (b) Series of pure-exchange CODEX spectra of unlabeled PMMA at 300 K,  $v_r = 6.5$  kHz, and  $Nt_r = 615 \ \mu s$ , as a function of  $t_m$  as indicated. For the COO carbon,  $\delta = 2\pi(91 \text{ ppm}) = 2\pi(6.9 \text{ kHz})$  so that  $\delta Nt_r = 8\pi$ ; for the OCH<sub>3</sub> group,  $\delta N t_r = 3.8\pi$ . Large-amplitude side group and smalleramplitude backbone motions are observed. Top: Full reference spectrum, scaled down 8 times. Experiments were performed at 75 MHz for <sup>13</sup>C in a 7-mm MAS probehead on a BRUKER DSX-300 spectrometer. <sup>13</sup>C 90° pulse length, 3.9  $\mu$ s; <sup>1</sup>H decoupling at  $\gamma B_1/2\pi = 70$  kHz; recycle delay, 1.5 s; CP time, 1.5 ms.; measuring time per PMMA spectrum, 2.5 h.

of Figure 1. Since the resulting difference spectrum is exclusively due to the exchanging sites, we term this pure-exchange CODEX NMR.

The smaller the motional amplitude, the longer the time  $Nt_r$ must be to produce significant dephasing. Taking into account the size  $\delta$  of the chemical-shift anisotropy,<sup>7</sup> the reorientation angle can be estimated from the dependence of the intensity on  $\delta N t_r$ (see Figure 4 below). In another series of experiments, the correlation time  $\tau_c$  is obtained as the time constant of the  $t_m$ dependent intensity change. The information content of the normalized exchange intensity  $E(t_m, \delta N t_r)$ , obtained as the ratio of the pure-exchange CODEX to the reference intensity, is similar to that of 1D PUREX<sup>10</sup> and stimulated-echo<sup>11,12</sup> data, which, however, are measured without sample spinning and thus have much inferior sensitivity and resolution. Information about the number M of equivalent orientational sites accessible in the motional process and the fraction  $f_{\rm m}$  of mobile segments is obtained from the final exchange intensity  $E_{\infty} = E(t_m \gg \tau_c, \delta N t_r \gg 1)$ 

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**Figure 3.** Normalized pure-exchange CODEX intensities  $E(t_m, \delta N t_r)$  as a function of  $t_m$  for (a) COO groups in methylmalonic acid (T = 300 K,  $\delta N t_r = 7.6\pi$ ), COO in PMMA (T = 288 K,  $\delta N t_r = 8\pi$ ), and CH<sub>3</sub> groups in DMS (T = 288 K,  $\delta N t_r = 12.4\pi$ ) and (b) CH and CH<sub>2</sub> groups in iPP (T = 365 K;  $\delta N t_r = 3.6\pi$  and  $8.5\pi$ , respectively, with  $\delta = 2\pi(11 \text{ ppm})$ for the CH groups). The fit curve for PMMA is a stretched exponential  $\exp(-(t_m/\tau_0)^{\beta})$  with  $\beta = 0.92$  and  $\tau_0 = 110$  ms; iPP is fit with  $\beta = 0.69$ and  $\tau_0 = 690$  ms.



**Figure 4.** Normalized pure-exchange CODEX intensities  $E(t_m, \delta N t_r)$  as a function of  $\delta N t_r$  for DMS at 288 K,  $t_m = 75$  ms, and  $\nu_r = 5.5$  kHz. Also shown are simulations for reorientations of the S–CH<sub>3</sub> axes by 10° or 170°, 20° or 160°, 30° or 150°, 50° or 130°, and 70° to 110° (curves in this large-amplitude range are virtually indistinguishable).

 $= f_{\rm m}(M-1)/M$ . For  $f_{\rm m} = 1$ , the minimum  $E_{\infty}$  is  $^{1}/_{2}$ , obtained for M = 2. Generally,  $E_{\infty} < f_{\rm m} \le 2E_{\infty}$ .

A reference virtually without exchange is provided by unlabeled methylmalonic acid. Figure 2a displays its CODEX spectrum for  $t_m = 200 \text{ ms}$  and  $\Delta = 1 \text{ ms}$ , the reference spectrum with  $\Delta$  and  $t_m$  interchanged, and their difference, the pure-exchange CODEX spectrum. The absence of intensity in the latter shows that no slow motions occur.

In contrast, large-amplitude motions of COOCH<sub>3</sub> sidegroups on a ~50-ms time scale at 300 K are easily observed by CODEX in amorphous poly(methyl methacrylate), PMMA (Plexiglas),  $[-CH_2-C(CH_3)((CO)-O-CH_3)-]_n$ , see Figure 2b. This motion is associated with the  $\beta$ -relaxation of PMMA,<sup>13</sup> which imparts reduced brittleness to this glassy polymer at ambient and elevated temperatures. Previously, detailed NMR studies of the  $\beta$ -relaxation dynamics in PMMA<sup>14,15</sup> required expensive <sup>13</sup>C and <sup>2</sup>H labeling combined with extensive multidimensional exchange NMR studies. It was concluded that ~50% of the sidegroups undergo 180° flips coupled to ~ $\pm 25^{\circ}$  rotations around the local chain axis. Using CODEX, this process can now be characterized in unlabeled industrial PMMA. Not only the sidegroup but also the backbone motions are detected directly in the pure-exchange CODEX spectra of Figure 2b.

In Figure 3, the  $t_{\rm m}$  dependence of the CODEX exchange intensity is plotted for four samples. For the carboxyl line of methylmalonic acid, only a few percent of exchange intensity is observed on a 1-s time scale, most likely due to <sup>13</sup>C spin diffusion. In dimethyl sulfone (DMS), (CH<sub>3</sub>)<sub>2</sub>SO<sub>2</sub>, the CH<sub>3</sub> axes reorient by 108° due to a 2-site jump process.<sup>16</sup> The CODEX data of Figure 3a show that the jumps of DMS indeed occur with M =2 ( $E_{\infty} = 0.50 \pm 0.04$ ) and  $\tau_c = 25$  ms. Isotactic polypropylene (iPP),  $[-CH_2-CH(CH_3)-]_n$ , is a semicrystalline polymer that can undergo jumps around the 31-helix axis in the crystalline regions.<sup>17,18</sup> The equilibration at  $E_{\infty} = 0.64$  (CH) and 0.67 (CH<sub>2</sub>) in Figure 3b confirms that the chains perform jumps between M= 3 NMR-distinguishable orientational sites. For PMMA at 288 K,  $E_{\infty}$  of only 0.23  $\pm$  0.02 in the exchange-intensity curve of Figure 3a shows that less than 46% of all sidegroups undergo large-amplitude motions. This estimate of the fraction  $f_m$  of flipping sidegroups from CODEX data is more precise than that obtained from static spectra in ref 14.

Estimates of the motional amplitudes can be obtained from the  $\delta N t_r$ -dependence, as shown for DMS in Figure 4. The simulations for different reorientation angles plotted for comparison exhibit a strong sensitivity to small-amplitude motions.

The efficiency of CODEX facilitates systematic studies of molecular dynamics as a function of temperature, processing, chemical modifications, etc. Various further developments of the CODEX technique are in progress.<sup>9</sup> With spin diffusion during  $t_{\rm m}$ , CODEX will also be useful for conformational analysis in <sup>13</sup>C-labeled materials. It is anticipated that the sensitive and relatively simple CODEX experiment will be applied routinely in solid-state NMR investigations of organic materials.

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